

Laboratorio di Elettrochimica  
dei Materiali per l'Energetica

# "Water processable polymers for supercapacitors and Li-ion batteries

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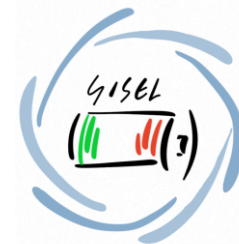
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LEME - Laboratory of Electrochemistry of Materials for Energetics

*Department of chemistry "Giacomo Ciamician"*

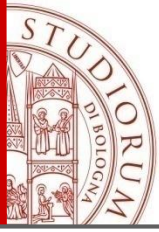
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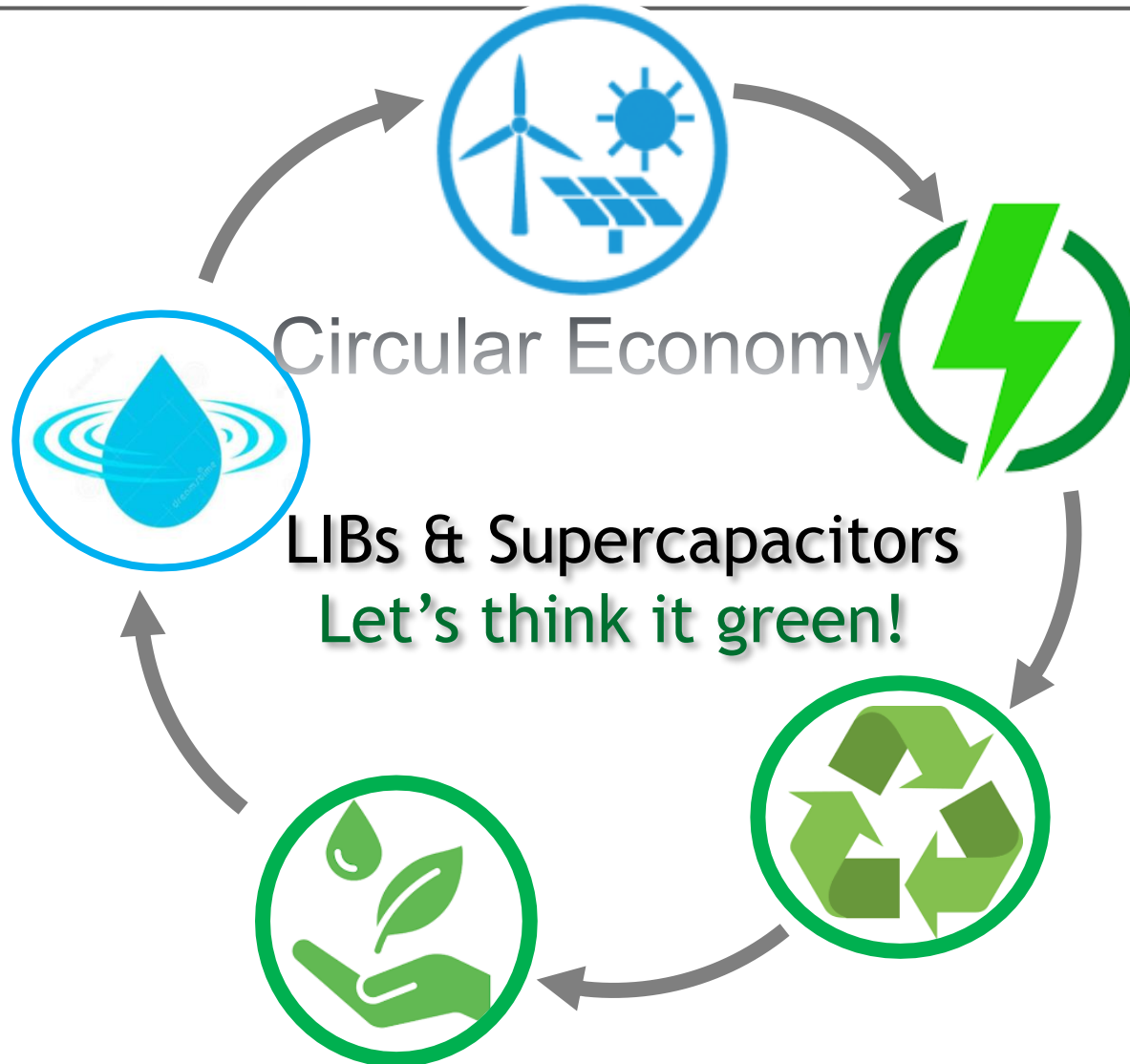


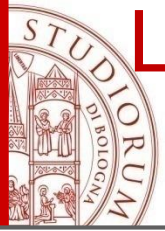
*Nanoinnovation 2020*

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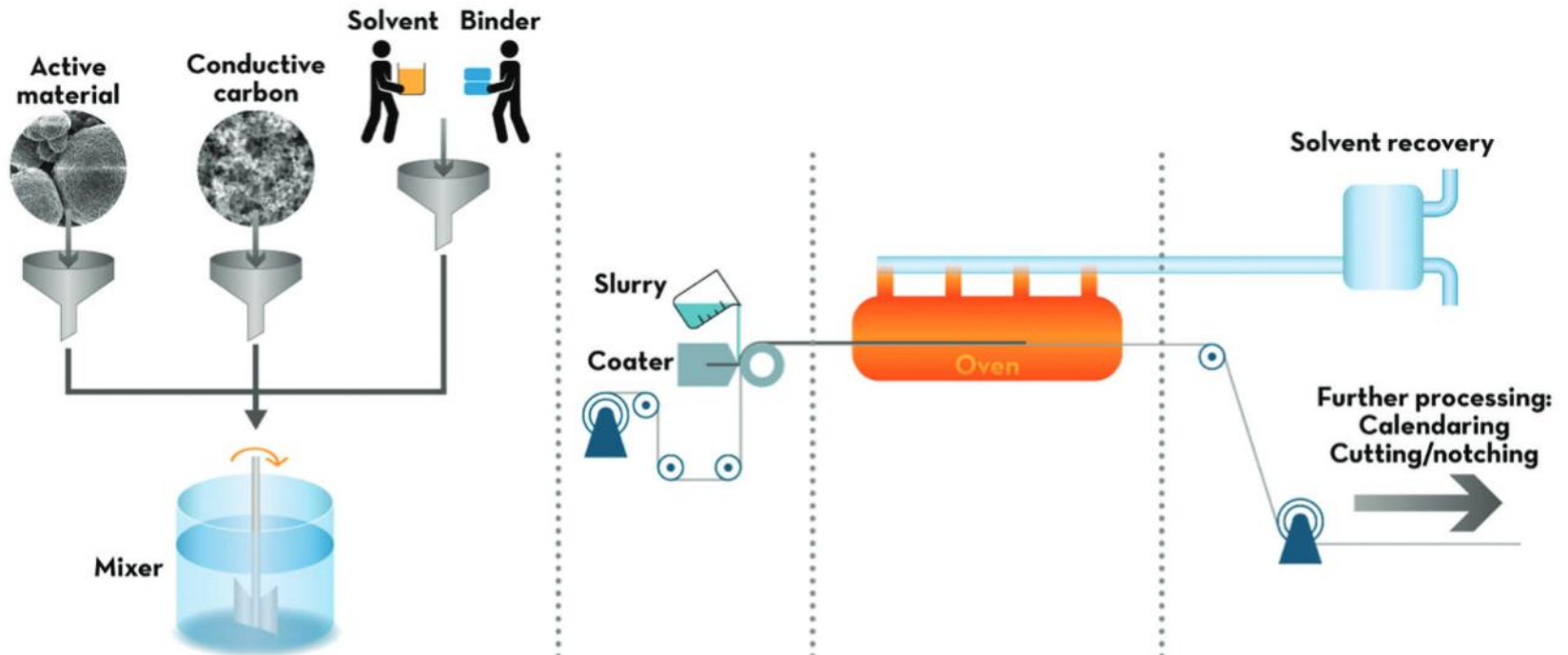


# EES in the Waste - Water – Energy Nexus

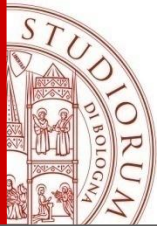




# LIB Electrode processing from NMP to Water processable binders



Solvent	Mixing	Coating	Drying	Recovery
<b>NMP</b>	Hazardous handling requires ventilation system and PSE	Controlled environment needed	Solvent vapours need to be collected to avoiding dispersion in the atmosphere	Recovery system required to recycle the expensive solvent
<b>Water</b>	PSE only required for handling of powders	Can be performed in air	No recovery necessary	Not needed



# Today EDLC ..not so green

**Separator:** cellulose, PTFE, PP ca. 20-50  $\mu\text{m}$ ,

**Electrolyte (organic):** *Tetraethylammonium tetrafluoroborate*  $\text{TEABF}_4$  in Acetonitrile,  
 $V_{\text{MAX}}=2.7 \text{ V}$

**Binder:** PTFE, PVDF, which require the use of *N-Methyl-2-pyrrolidone (NMP)* to be processed

**Acetonitrile**



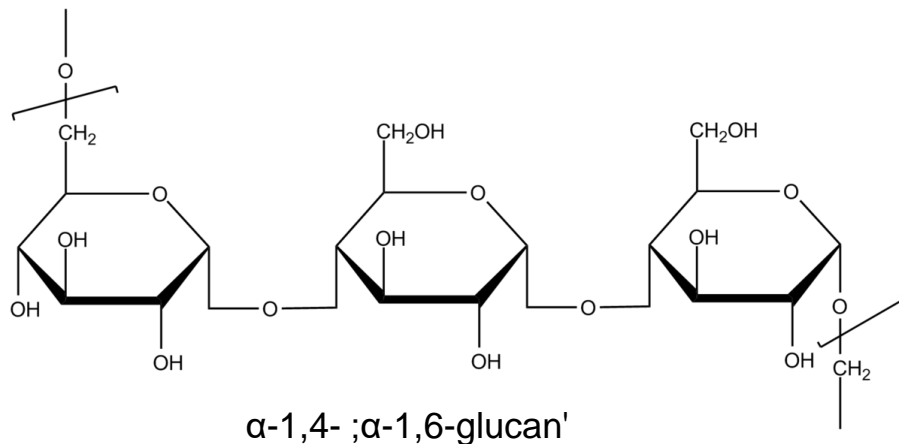
**N-metil pirrolidone (NMP)**



***Tetraethylammonium tetrafluoroborate***



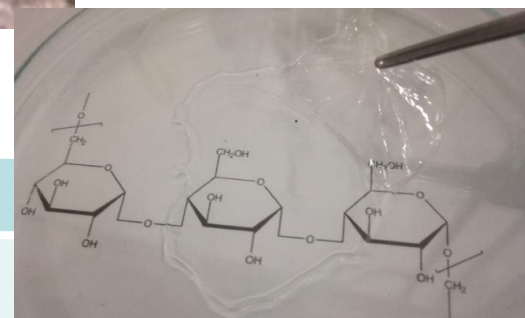
# Pullulan for EES



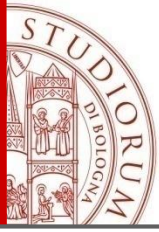
*Aureobasidium pullulan*  
fungus



Epiphyte



Chemistry	Film properties	Role in food packaging
Polysaccharide (maltotriose)	<ul style="list-style-type: none"> <li>-Biodegradable</li> <li>-Transparent</li> <li>-Edible</li> <li>-Oil and grease resistant</li> <li>-Heat sealable</li> <li>-High water solubility</li> <li>-Barrier to oxygen</li> <li>-Stable in a wide pH range</li> </ul>	<ul style="list-style-type: none"> <li>-Coating/wrapping</li> <li>-Blend with other polymers to improve mechanical properties</li> <li>-Inner package</li> </ul>



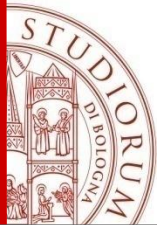
# Pullulan for EES

Electrolyte affinity, electrochemical & chemical stability:

- Electrospun Pullan separator (Supercap)

Binding properties:

- Thick porous carbon electrodes (Supercap)
- High voltage LIB cathodes



# Pullulan-based separator

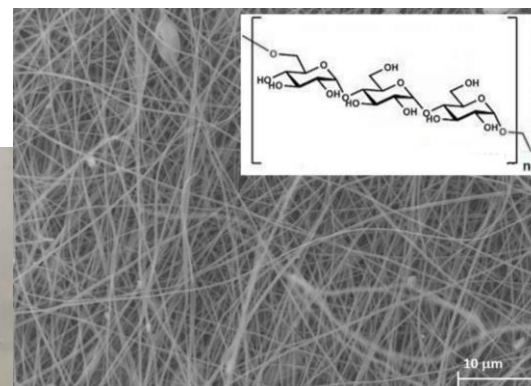
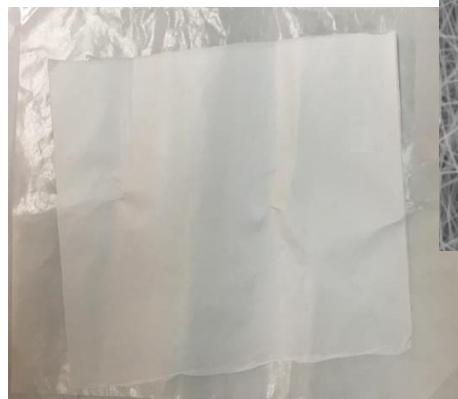
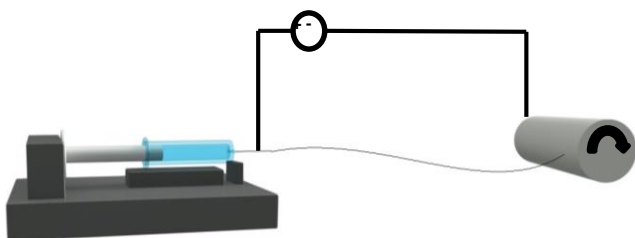


## Pullulan mats prepared by electrospinning

from a 23% w/v solution of pullulan in Milli-Q water.

Spun at 18 kV at 20 cm from the collector with a flow rate of 1 mL h<sup>-1</sup>.

Non woven pullulan mats thickness of 55  $\mu\text{m}$  with average fiber diameter of 0.30  $\mu\text{m}$

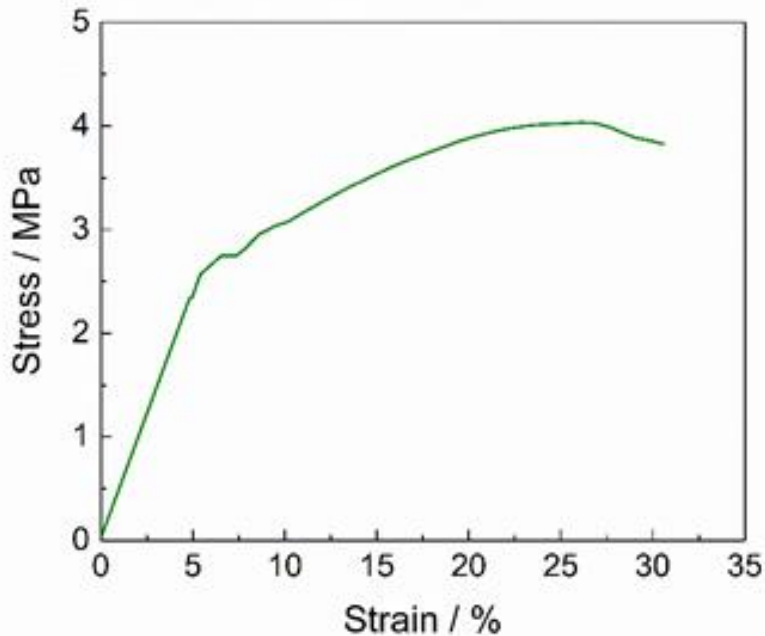


SEM at 200 kX



# Pullulan electrospun separator

## Tensile Stress strain

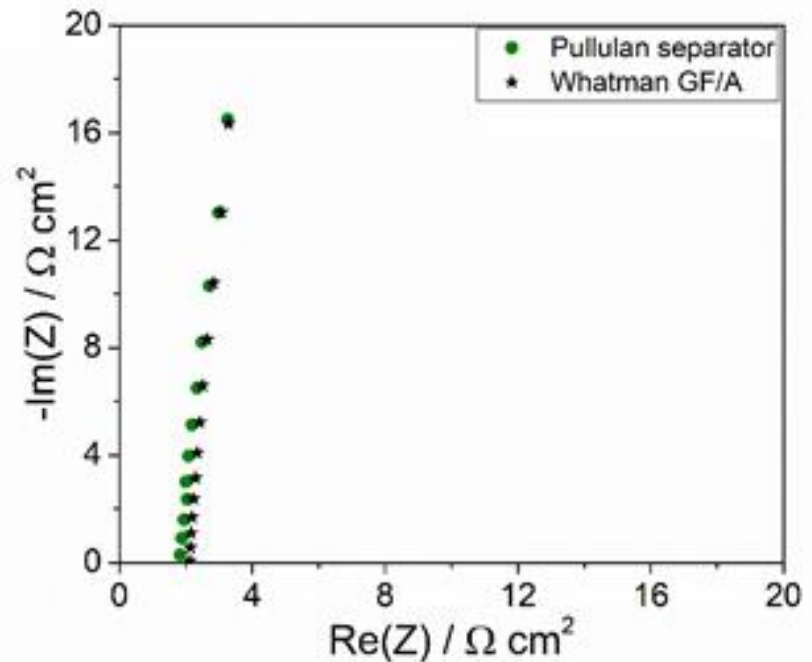


Mat elastic modulus  $E = 85 \pm 27$  Mpa

Stress at break  $s_b = 3.4 \pm 0.4$  MPa

Strain at break  $e_b = 32 \pm 9$  %.

## Electrochemical impedance spectroscopy

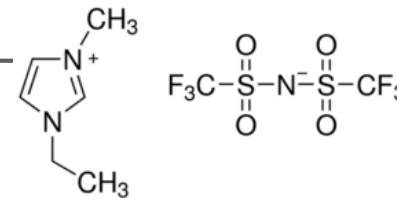


**Symmetric cells with stainless steel blocking electrodes separated by the Pullulan membrane and Whatman GF/A, (fibre glass) soaked in EMIMTFSI (frequency range from 500 kHz to 15 kHz)**



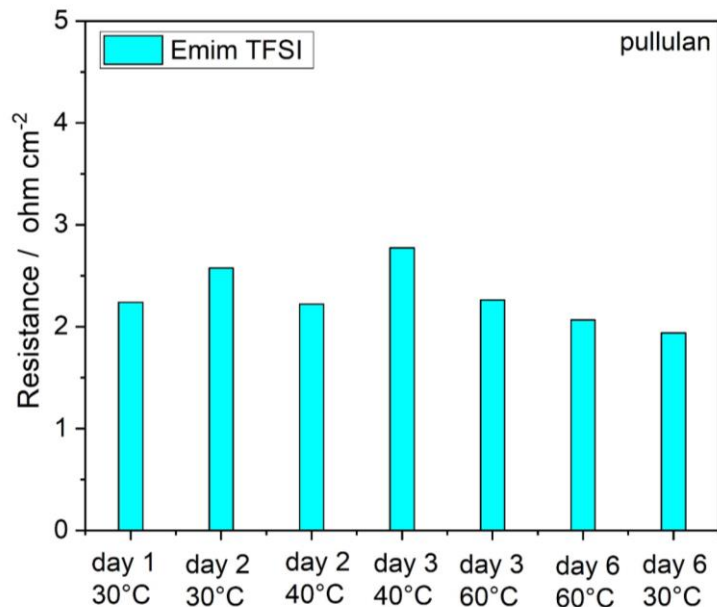
# Pullulan-based separator in IL

From EIS with stainless steel blocking electrodes separated by the PU mat soaked with EmimTFSI

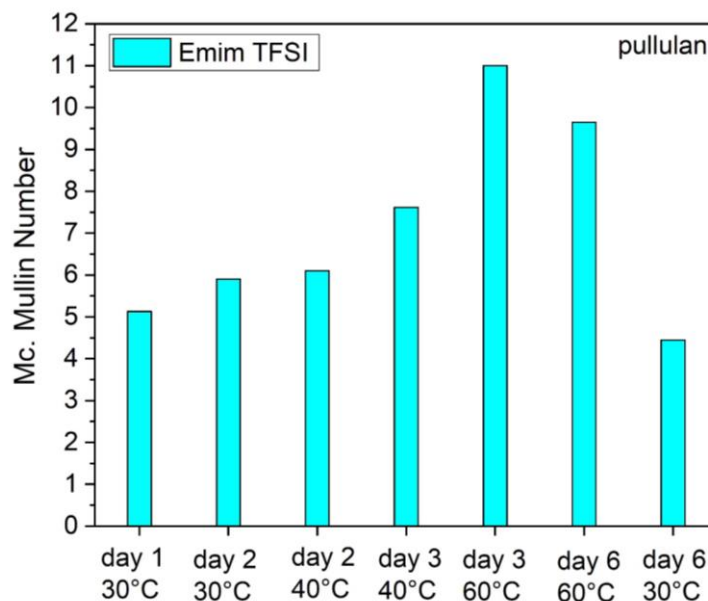


$N_M = \rho_{eff} / \rho_0$  resistivity of the separator soaked in the electrolyte ( $\rho_{eff}$ ) with respect to the bulk resistivity of the electrolyte solution ( $\rho_0$ )

## PU-EmimTFSI ionic Resistance



## PU MacMullin number ( $N_M$ )



The PU mat showed low values of resistivity and MC number, constant over time

# Pullulan as Electrode Binder for Supercapacitors

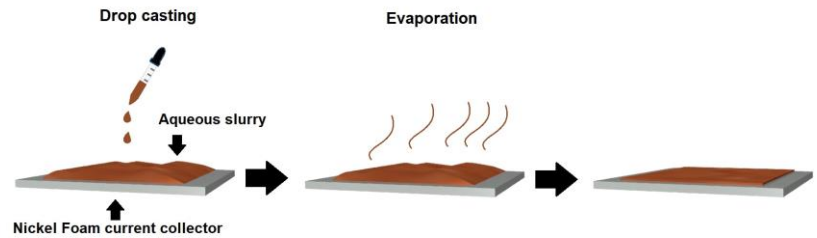
Evaluation of the binding efficiency in High Binder Low Mass Electrode (HBLME) and Low Binder High Mass Electrode (LBHME).

Commercial mesoporous carbon PICACTION from PICA (BP10, BET specific surface area  $1900 \text{ m}^2 \text{ g}^{-1}$ )

Binder: Pullulan:Glycerol 1:1

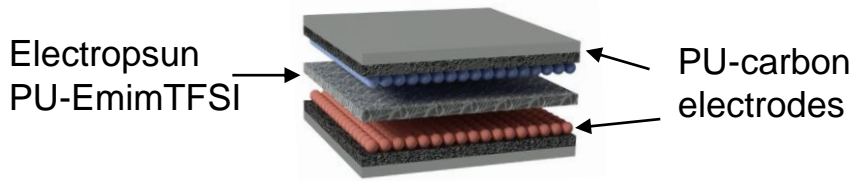
Name	Composition	Mass loading range*
HBLME	70%BP10 / 10%CB / 20%binder	3.6-4.6 $\text{mg cm}^{-2}$
LBHME	85%BP10 / 5%CB / 10%binder	6.3-7.5 $\text{mg cm}^{-2}$

\*single electrodes mass loading



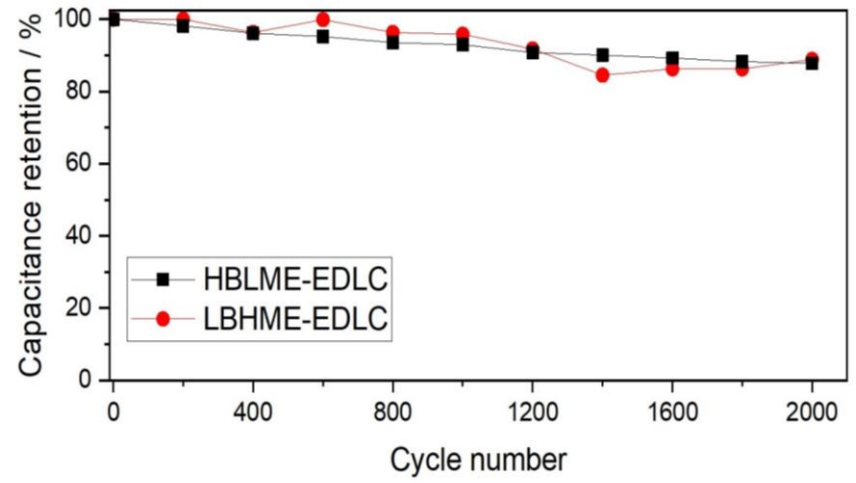
Schematic illustration of drop casting process.

## PU-Supercapacitor



**PU is an excellent binder even for thick electrodes working at high potentials**

Supercapacitor GLV cycling @ 1 A/g between 0 and 3.2 V



# Pullulan-based Supercapacitors

Commercial mesoporous from agriculture waste (biochar)

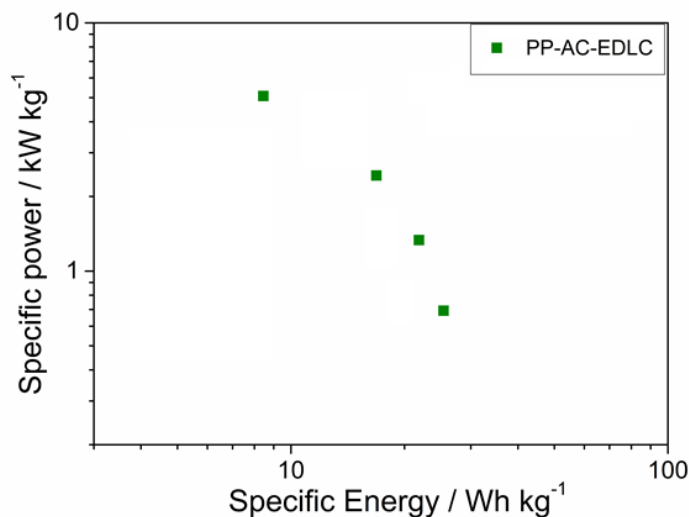
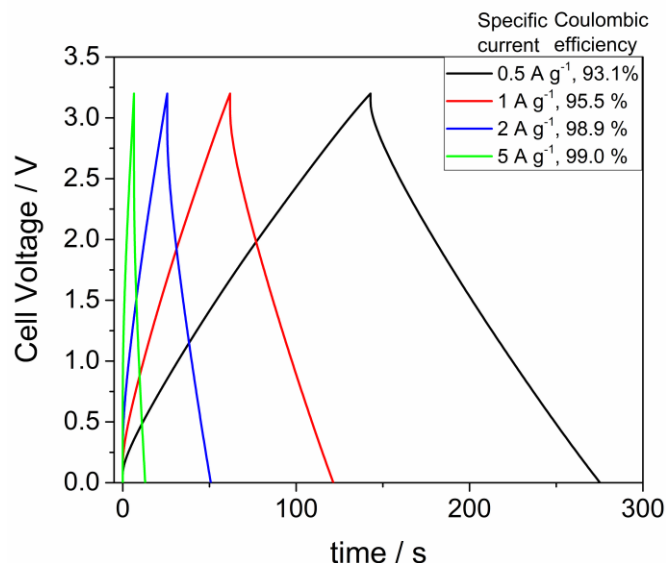
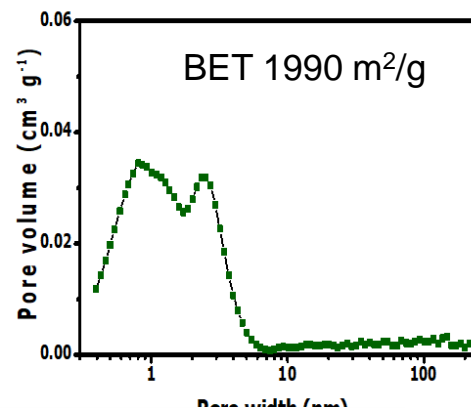


Carbon derived from the pyrolysis of bell peppers seeds at 850 °C with  $\text{KHCO}_3$  activating agent

Electrode mass loading 6 - 10 mg  $\text{cm}^{-2}$

Electrode composition: 70% AC-10 % acetylene black- 20 % PU

Electrolyte: EmimTFSI – Separator: Electrospun PU



Specific capacitance 27  $\text{F g}^{-1}$   
 Specific power 7  $\text{kW kg}^{-1}$  at 5  $\text{A g}^{-1}$   
 Specific energy 25  $\text{Wh kg}^{-1}$  at 0.25  $\text{A g}^{-1}$

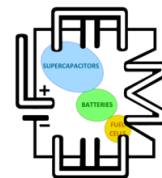
Stable over more than 2000 cycles

# Conclusions

- Pullulan can be used to design the major components (binder and separator) of green supercapacitors by sustainable, water based manufacturing procedures (electrospinning).
- The electrode and membrane processes that we propose represent environmentally improved and safer routes that can substitute the conventional manufacturing of fluorinated polymers based on the use of the toxic solvent N-Methyl-pyrrolidone.
- Pullulan can be exploited for the aqueous processing of NMC cathodes with comparable performance to PVdF-based electrodes
- LIB cathode preparation procedure should be further optimized to achieve full exploitation of the active material



# Acknowledgements



• “Realizzazione di catodi per batterie con anodi di litio metallico ”  
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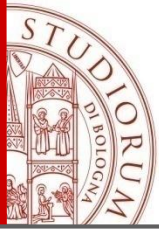
*LEME – indoor*



*LEME – outdoor*







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